

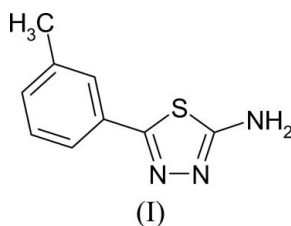
5-*m*-Tolyl-1,3,4-thiadiazol-2-ylamineFeng Han, Rong Wan,*
Wen-Yuan Wu, Jin-Jun Zhang
and Jin-Tang WangDepartment of Applied Chemistry, College of
Science, Nanjing University of Technology, No.
5 Ximofan Road, Nanjing, Nanjing 210009,
People's Republic of China

Correspondence e-mail: rwan01@jlonline.com

Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.006$ Å
 R factor = 0.063
 wR factor = 0.150
Data-to-parameter ratio = 15.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.The asymmetric unit of the title compound, $\text{C}_9\text{H}_9\text{N}_3\text{S}$, contains four molecules which are interconnected through $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonding, resulting in the formation of a tetrameric structure.Received 6 December 2006
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Comment

1,3,4-Thiadiazole derivatives represent an interesting class of compounds possessing a broad spectrum of biological activities (Nakagawa *et al.*, 1996; Wang *et al.*, 1999). These compounds are known to exhibit diverse biological effects, such as insecticidal and fungicidal activities (Wang *et al.*, 1999).The asymmetric unit of the title compound, (I), is built up from four independent molecules (*A*, *B*, *C* and *D*) interconnected through $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonding, resulting in the formation of a pseudo-tetrameric structure (Fig. 1). The four molecules are roughly planar. The thiadiazole and tolyl rings are only slightly twisted with dihedral angles of 5.7° (*A*), 10.8° (*B*), 16.5° (*C*) and 3.4° (*D*).

Experimental

3-Methylbenzoic acid (5 mmol) and thiosemicarbazide (5 mmol) were added to toluene (50 ml) and heated under reflux for 4 h. The reaction mixture was left to cool to room temperature and then filtered; the filter cake was crystallized from acetone to give pure compound (I) (m.p. 513–514 K). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an acetone solution.

Crystal data

 $\text{C}_9\text{H}_9\text{N}_3\text{S}$
 $M_r = 191.26$
Triclinic, $P\bar{1}$
 $a = 10.793$ (2) Å
 $b = 10.820$ (2) Å
 $c = 17.607$ (4) Å
 $\alpha = 73.85$ (3) $^\circ$
 $\beta = 78.97$ (3) $^\circ$
 $\gamma = 73.51$ (3) $^\circ$ $V = 1879.6$ (8) Å³
 $Z = 8$
 $D_x = 1.352$ Mg m⁻³
Mo $K\alpha$ radiation
 $\mu = 0.30$ mm⁻¹
 $T = 293$ (2) K
Block, colorless
 $0.30 \times 0.20 \times 0.20$ mm

Data collection

Enraf–Nonius CAD-4
diffractometer
 $\omega/2\theta$ scans
Absorption correction: ψ scan
(North *et al.*, 1968)
 $T_{\min} = 0.924$, $T_{\max} = 0.943$
7785 measured reflections

7370 independent reflections
4158 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.024$
 $\theta_{\max} = 26.0^\circ$
3 standard reflections
every 200 reflections
intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.063$
 $wR(F^2) = 0.150$
 $S = 1.02$
7370 reflections
473 parameters
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.053P)^2 + 0.39P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.22 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.26 \text{ e } \text{Å}^{-3}$

Table 1

Hydrogen-bond geometry ($\text{Å}, ^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N3C-H3C1 \cdots N2B$	0.86	2.17	2.994 (5)	161
$N3B-H3B1 \cdots N2C$	0.86	2.14	2.959 (5)	161
$N3A-H3A2 \cdots N1B$	0.86	2.42	3.187 (4)	149
$N3A-H3A2 \cdots N2B$	0.86	2.33	3.169 (5)	164
$N3D-H3D2 \cdots N1C$	0.86	2.22	3.064 (4)	169
$N3D-H3D1 \cdots N2D^i$	0.86	2.22	3.066 (4)	169
$N3C-H3C2 \cdots N1D^{ii}$	0.86	2.18	3.004 (4)	161
$N3A-H3A1 \cdots N2A^{iii}$	0.86	2.21	3.014 (5)	155

Symmetry codes: (i) $-x + 1, -y, -z + 1$; (ii) $x + 1, y, z$; (iii) $-x + 2, -y + 1, -z$.

All H atoms were positioned geometrically and treated as riding on their parent atoms, with $C-H = 0.93$ (aromatic) or 0.96 Å (methyl), $N-H = 0.86 \text{ Å}$, and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{aromatic C, N})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

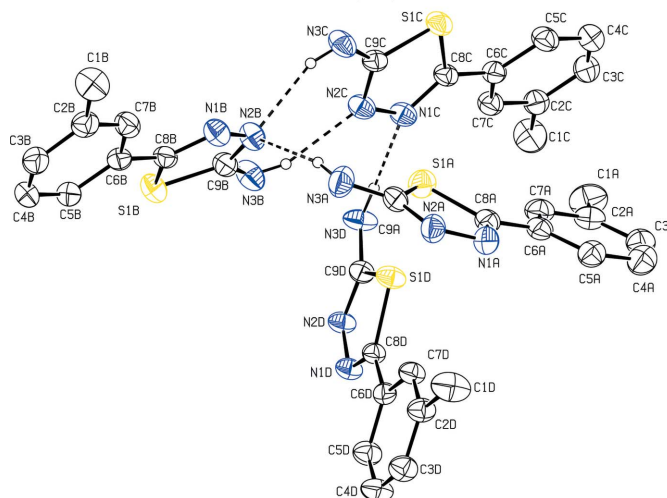


Figure 1

The asymmetric unit of (I). Displacement ellipsoids are drawn at the 30% probability level. H atoms are represented as small spheres of arbitrary radius. $N-H \cdots N$ bonds are represented as dashed lines. H atoms not involved in hydrogen bonding have been omitted for clarity.

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